

## CHMT (Chemical Technology)

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### CHMT 1120 - Chemical Technology I

(6 Credits)

This course is an introduction to basic concepts in chemistry. The course covers chemical properties, identification and separation of substances, atomic and molecular structure, elements and compounds, liquids and solutions; the periodic table and the naming of inorganic substances. Students will use a variety of instruments including the gas chromatograph, manometers, analytical balances and visible spectrophotometers. Lecture: 2 hours, Lab: 8 hours - Lab Fee: \$20

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### CHMT 1121 - Chemistry for Biotechnology

(3 Credits)

This course provides an introduction to basic concepts in chemistry and instruction in the use of instruments, including gas and liquid chromatographs, electronic balance, visible spectrophotometers and pH meters. Topics covered include chemical properties, identification and separation of substances, atomic structure, elements and compounds, gases and solutions, acids and bases and amino acids and proteins. Lecture: 2 hours, Lab: 3 hours - Lab Fee: \$20

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### CHMT 1220 - Chemical Technology II

(6 Credits)

This course is a continuation of Chemical Technology I and covers acid-base chemistry, equilibrium, qualitative and quantitative analysis. (Prerequisite: CHMT 1120 or its equivalent) Lecture: 2 hours, Lab: 8 hours - Lab Fee: \$20

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### CHMT 2320 - Chemical Technology III

(10 Credits)

This course is a continuation of Chemical Technology II and covers an introduction to organic chemistry via a study of organic functional groups, classification of organic compounds using wet and instrumental methods and organic reactions. It will also focus on infrared spectrophotometry separation methods, including gas chromatography and high pressure liquid chromatography (Prerequisite: CHMT 1220 or its equivalent) Lecture: 4 hours, Lab: 12 hours - Lab Fee: \$20

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### CHMT 2420 - Chemical Technology IV

(8 Credits)

This course is a continuation of Chemical Technology III and covers the nature of reversible processes, equilibrium constants, solute and solvent systems and the kinetics of chemical reactions. Instrumental methods used include atomic absorption, emission spectroscopy, ultraviolet and NMR spectroscopy. The course concludes with a variety of special methods and student projects. (Prerequisite: CHMT 2320 or its equivalent) Lecture: 4 hours, Lab: 8 hours - Lab Fee: \$20